Topic # 6 ATMOSPHERIC STRUCTURE & CHEMICAL COMPOSITION Part II

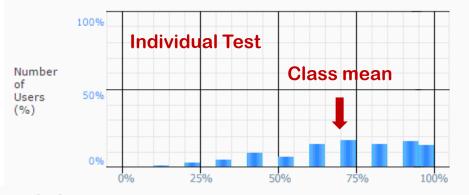
Plus wrap up of some other topics

Individual Test #1 Class Statistics

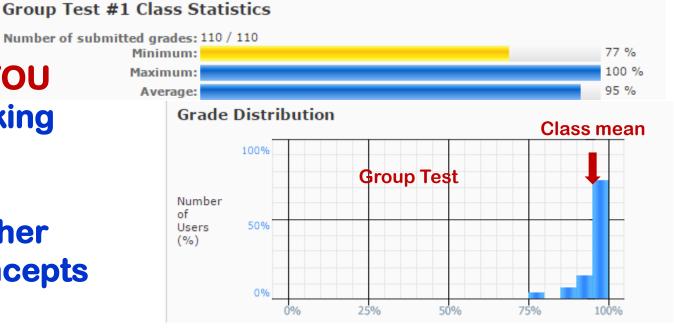


THE POWER OF COLLABORATIVE LEARNING!

Grade Distribution



REMEMBER: Your group is there to HELP YOU LEARN by working out problems together that & helping each other understand concepts



ABOUT THE LEARNING PHILOSOPHY UNDERLYING THIS COURSE

FAQ #2: <u>http://www.ltrr.arizona.edu/kkh/natsgc/faq.htm#2</u>. This course is designed to help you learn the course material in ways that are different from traditional lecture courses. . . .

<u>Class time is valuable:</u> Class should be active w/ hands-on activities so that <u>students can be **engaged in their own learning**</u>

• <u>Students learn best</u> when they **come to class prepared** (by reading, Self tests &RQs) to which new material is added

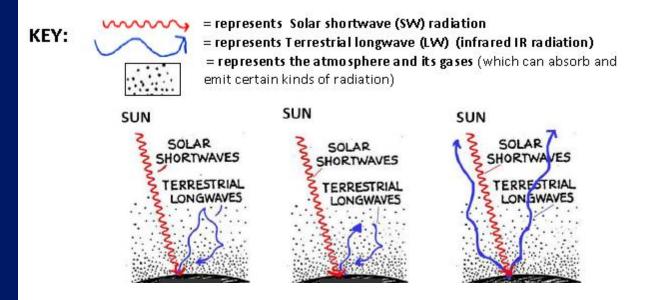
• <u>Students learn best</u> when they have a common "starting place" for deepening their understanding (why Dr H reviews concepts)

• <u>A group of people working together cooperatively can solve</u> complex problems more easily and at a higher level than an individual working alone

• <u>To achieve success in group activities</u>, each individual has to be perceived as someone with something to offer, and each individual is responsible for preparing and contributing his or her own individual knowledge.

GROUP WORK SELF TEST

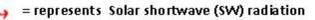
SOLAR vs TERRESTRIAL RADIATION CLASS CONCEPTS SELF TEST



5 minute GROUP WORK: Answer Q7 – Q11

ANSWERS

SOLAR vs TERRESTRIAL RADIATION CLASS CONCEPTS SELF TEST

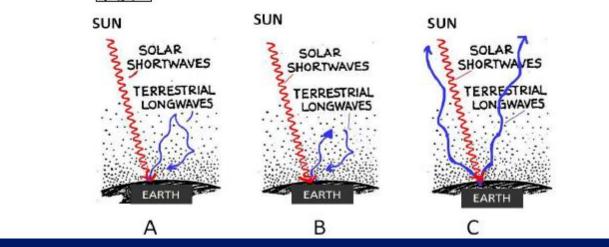


emit certain kinds of radiation)

KEY:

= represents Terrestrial longwave (LW) (infrared IR radiation)

= represents the atmosphere and its gases (which can absorb and



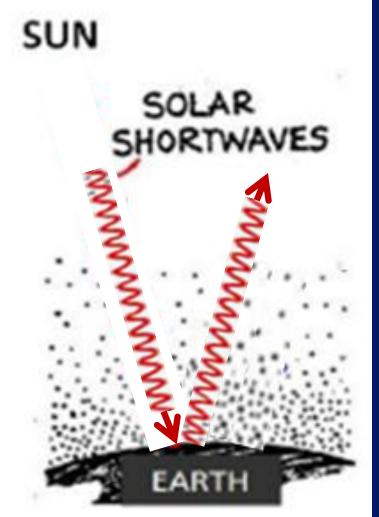
All 3 are illustrating <u>ABSORPTION</u> of incoming Solar SW by the EARTH's surface followed by <u>outgoing RADIATION</u> of LW Infrared from the EARTH's surface !

Q7. Which diagram above shows SW (solar radiation being <u>reflected</u> back to space?

B

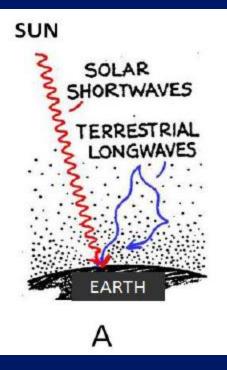


None of them



NO ABSORPTION

Here's the correct diagram to show SW Solar being <u>reflected</u> back to space!

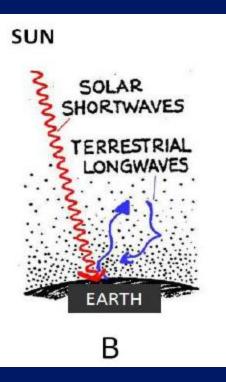


Q8. Diagram A shows LW (IR) terrestrial radiation "bouncing off" (or reflecting) the gases in the atmosphere and being sent back to Earth's surface. (i.e. being <u>reflected back</u> to the surface by the gases <u>without being absorbed</u> by them.)

Is this an accurate depiction of how the Greenhouse Effect works? Yes No Why or Why not?

The LW (IR) radiation is absorbed by the GHG's and then re-radiated (or re-emitted) back down to the Earth's surface to warm it. The IR is NOT reflected. IF IT'S REFLECTED IS NOT ABSORBED.

DON'T USE "BOUNCING or "REFLECTING" to describe the Greenhouse Effect process: GH gases <u>ABSORB</u> & <u>RE-RADIATE</u>!



Q9. Diagram B shows LW (IR) terrestrial radiation being <u>absorbed</u> <u>and then emitted back down</u> by the gases in the atmosphere.

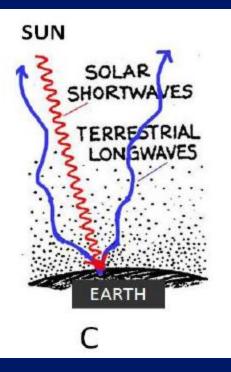
Is this an accurate depiction of how the Greenhouse Effect works?

Yes

Why or Why not?

(although it's not quite complete – more on this later)

LW (IR) being absorbed by GH Gases, and then emitted out again, (back down to the surface of the Earth) is exactly how the GH Effect operates.



Q10. Diagram C shows LW (IR) terrestrial radiation going right through the atmosphere <u>out to</u> <u>space.</u>

Is this an accurate depiction of how the Greenhouse Effect works?

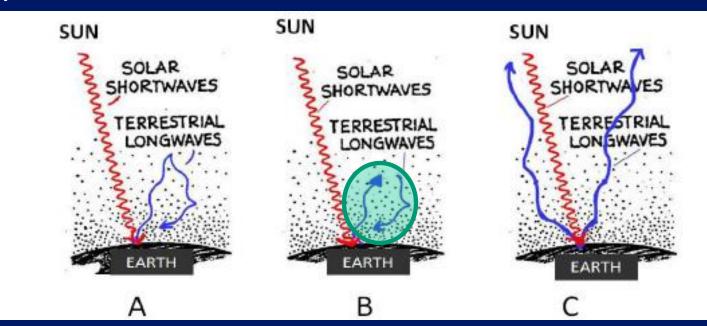
Yes No

Why or Why not?

Diagram C shows <u>ALL</u> the IR leaving the Earth's surface and <u>NOT</u> being absorbed at all !

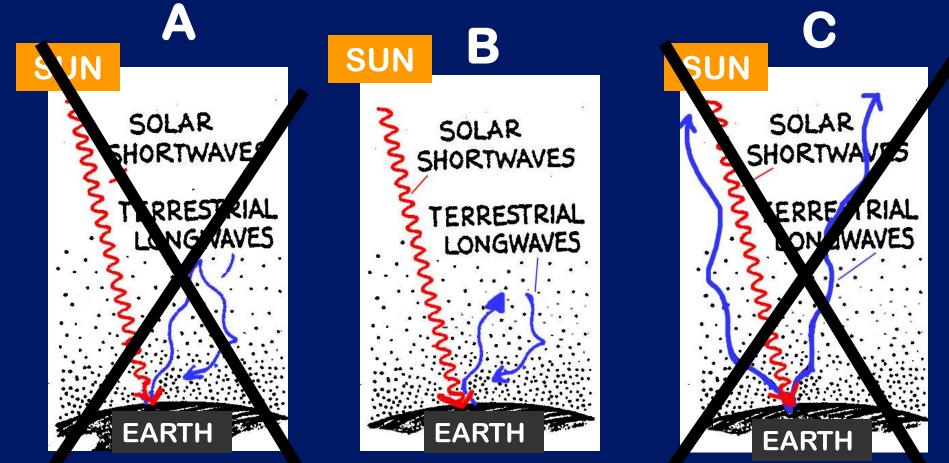
If this happened the Earth's temperature would be below freezing!

Q11. On the diagram that you think best depicts the processes involved in the GREENHOUSE EFFECT, <u>CIRCLE</u> the <u>specific part</u> of the diagram that represents the Greenhouse Effect:

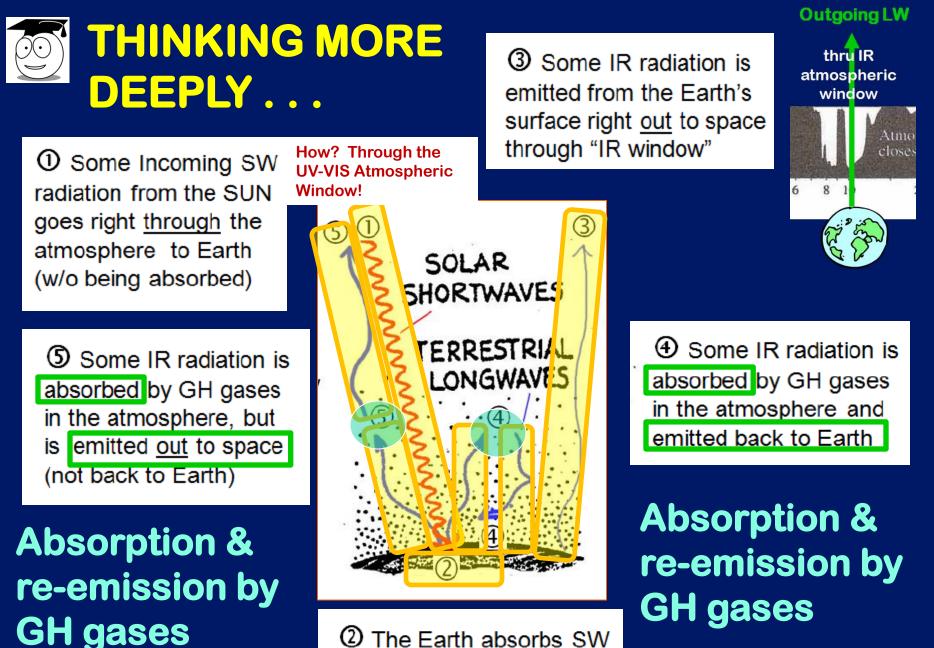


DO <u>NOT</u> CIRCLE any part of the SUN's incoming SW! (Shortwave Radiation = IR + VIS) SW is NOT part of the GH Effect!

Most accurate depiction of the GH Effect?

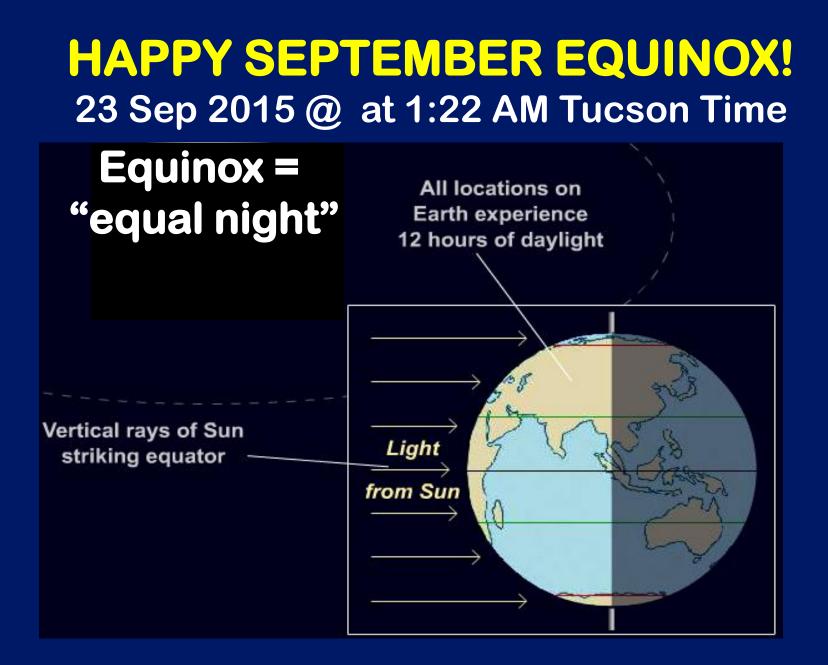


B is close, but actually, NONE of these is <u>exactly</u> correct . . . for a better version: see the bottom of p 29 in Class Notes.

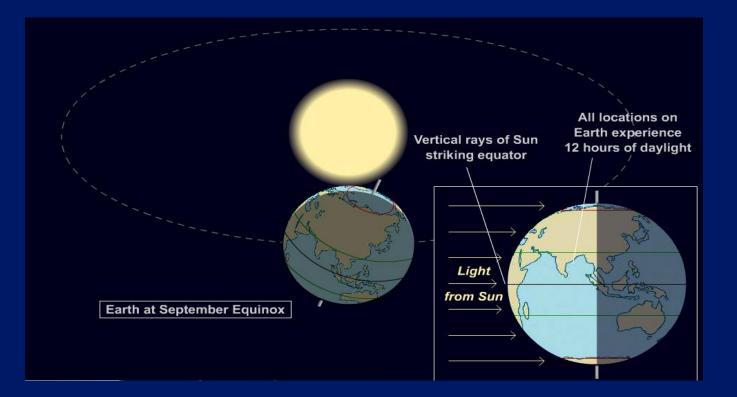


② The Earth absorbs SW that reaches the surface

Bottom of p 29



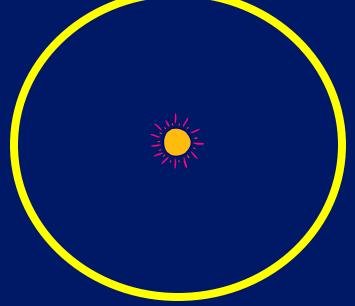
Today is the "Astronomical" "First Day of Fall" in the Northern Hemisphere!



You can view the animation yourself at:

http://mesoscale.agron.iastate.edu/agron206/animations/01 EarthSun.html





(1) Draw a circular ORBIT on the whiteboard & show SUN's location then use a phone flashlight to simulate Sun's radiation



(1) Draw a circular ORBIT on the whiteboard & show SUN's location then use a phone flashlight to simulate Sun's radiation

(2) Then, keeping the Earth's axis at a constant 23 ½ angle, and keeping the "north pole" axis 'tip' always pointing to the "North Star" – simulate a year of the Earth's Orbit – and discuss how the amount of radiation varies at different latitudes → seasonal changes!

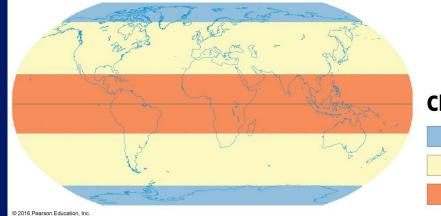


BUT . . . the Earth's orbit at this point in time is NOT circular, it's ELLIPTICAL & the SUN is NOT in the center!

How does this change the amount of Solar radiation received in different hemispheres?

At what latitudes is this difference the most drastic?

In the far north energy from the Sun is dispersed.



Climatic bands

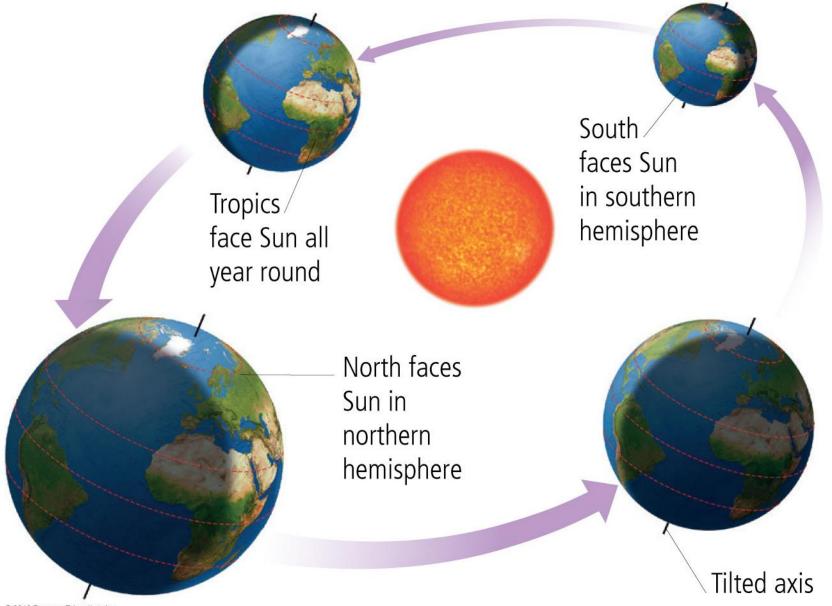
Polar regions
Temperate zones
The tropics

rctic Ocean Atlantic Tropic of Cancer Tropic of Cancer Ocean Pacific Ocean Equato Equato Pacific Ocean Indian Ocean Tropic of Capricorn Capricorr Antarctic Circle Antarctic Circle **Climatic zones Ocean currents** Warm Ice cap Tundra Humid equatorial Subarctic Continental Warm temperate Semi-aric Hot humic Tropical Cold © 2016 Pearson Education, Inc.

READ: Dire Predictions pp 10-15

In the tropics energy from the Sun is concentrated.

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READ: Dire Predictions pp 10-15

We'll address more on EARTH-SUN relationships , how they've <u>changed in the past</u>, and how this is an important <u>NATURAL</u> cause of <u>CLIMATE CHANGE</u>

in Topics #10 & #11



THINK for 15 seconds TABLE CHAT for 15 seconds What's your most burning question about: EARTH-SUN RELATIONSHIPS , the SEASONS and GLOBAL CLIMATE PATTERNS?

Topic # 6 ATMOSPHERIC STRUCTURE 8 CHEMICAL COMPOSITION All about the GASES IN THE **ATMOSPHERE**, esp. **GREENHOUSE GASES!**

OBJECTIVES:

To understand:

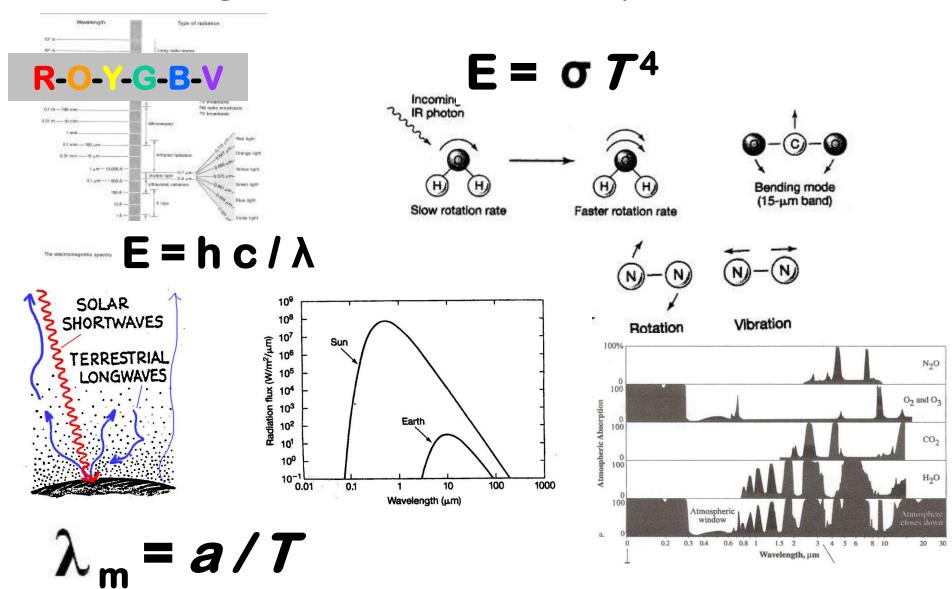
-- the VERTICALSTRUCTURE of the atmosphere & its relationship to temperature

-- which GASES are in the atmosphere

-- where they are concentrated, and

-- why gases at different levels are linked to the Greenhouse Effect and also Ozone Depletion (which are NOT the same thing!!!)

Things you've seen before that will all come together under this topic:



We travel together, passengers in a little space-ship, dependent on its vulnerable supplies of air and soil.

~ Adlai Stevenson



The atmosphere has a "structure" of different named layers :

{- 320km} (195.6{mi)} -Thermosphere

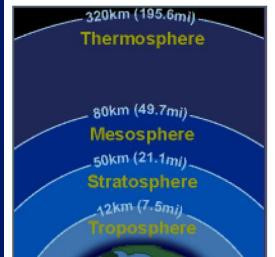
_ 80km (49.7*mi*). Mesosphere

50km (21.1mi)

Stratosphere

12km (7.5mi)

Troposphere



Mesopause Boundary between the mesosphere and the thermosphere. _

Stratopause

Boundary between the stratosphere and the mesosphere.

Atmospheric ozone layer Layer within the stratosphere. Absorbs ultraviolet solar radiation, so warming the surrounding atmosphere.

Tropopause

Boundary between the troposphere and the stratosphere. _



Exosphere

Outermost layer of the atmosphere. Extends to about 10,000 km (6,000 miles).

Thermosphere Extends to about 640 km (400 miles).

Mesosphere

Rises from about 50 to 80 km (30 to 50 miles) above the surface. Air becomes cooler as the altitude increases.

Stratosphere

Extends upward to a height of about 50 km (30 miles). Contains atmospheric ozone layer. Temperature increases with altitude through the stratosphere, inhibiting vertical air currents, and making the stratosphere highly stable, in contrast to the troposphere.

Troposphere

Layer in contact with Earth's surface. Extends from the surface to about 8 to 17 km (5 to 10 miles). Air temperature decreases with altitude, leading to instability. Less dense air sits below more dense air, which results in air movements and storm generation. "Weather" takes place almost exclusively within the troposphere.

Sea level

Dire Predictions p 12

MAKE A GRAPH:



ALTITUDE Kilometers (km)

100

0

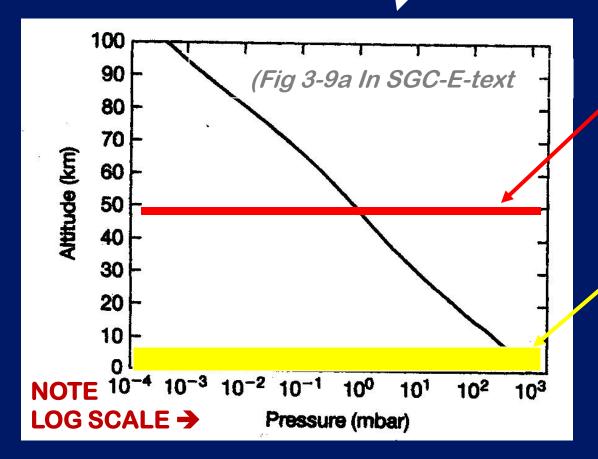
above sea level)

Low ATMOSPHERIC PRESSURE

High

Atmospheric Pressure = weight of the air column above

Atmospheric Pressure & Mass Vary with Height



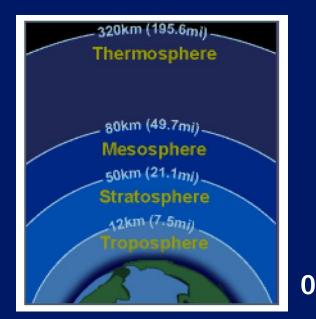
99% of mass lies below ~ 50 km (top of Stratosphere) 50% of mass lies below ~ 6 km (middle **Troposphere**)

MAKE A GRAPH:



ALTITUDE Kilometers (km) above sea level)

100



Thermosphere80 km – 320 kmMesosphere50 km – 80 kmStratosphere12 km – 50 kmTroposphere0 km – 12 km

Now erase pressure and sketch in the layers on your graph with a dashed line -----

MAKE A GRAPH:



ALTITUDE Kilometers (km) above sea level)

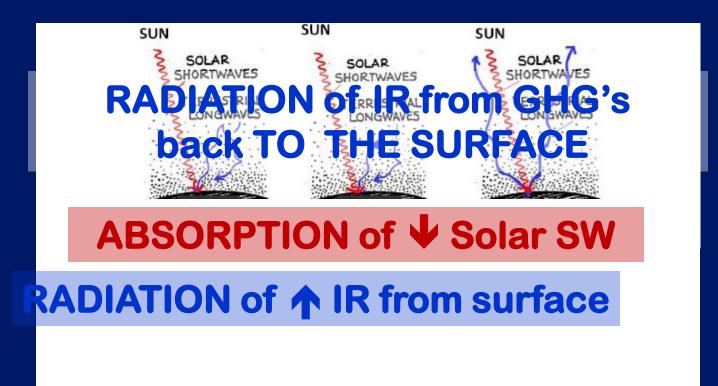
100

Sketch in Temperature curve

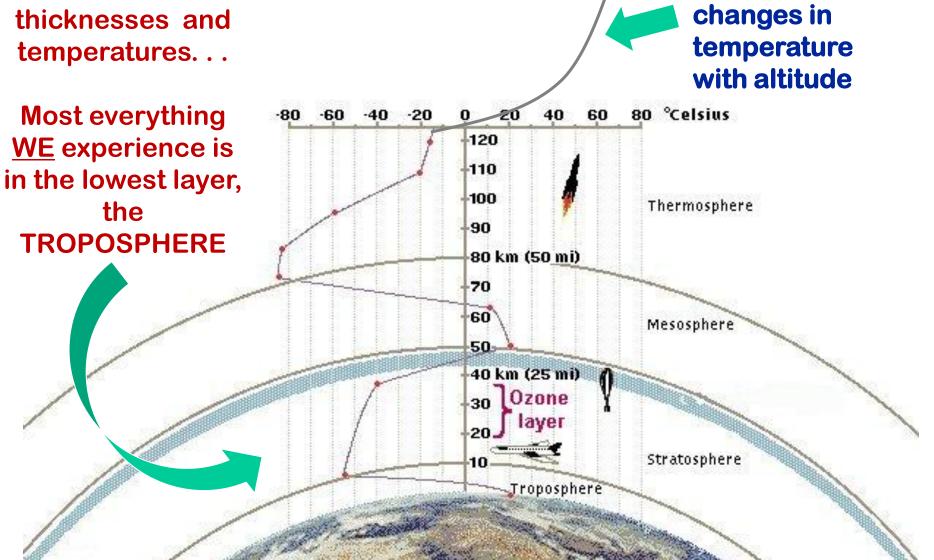
0 ____ COLD

ATMOSPHERIC TEMPERATURE HOT

The ATMOSPHERE IS PRIMARILY HEATED FROM BELOW (by IR energy radiated up from the EARTH'S SURFACE)



These layers have different thicknesses and temperatures...



This zig-zag line

is showing

The Vertical Structure of the Atmosphere

KEY CONCEPT:

The atmosphere's vertical structure is defined by CHANGES in the trend of TEMPERATURE with height.

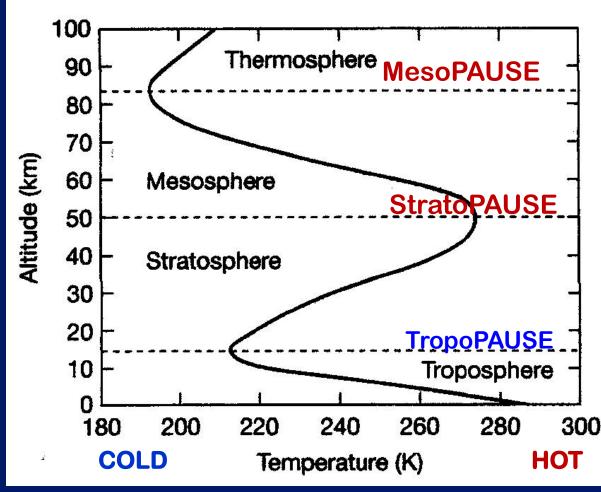
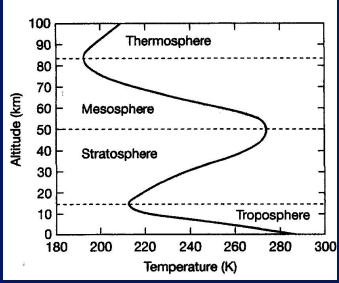


Figure 3-9b in SGC E-text



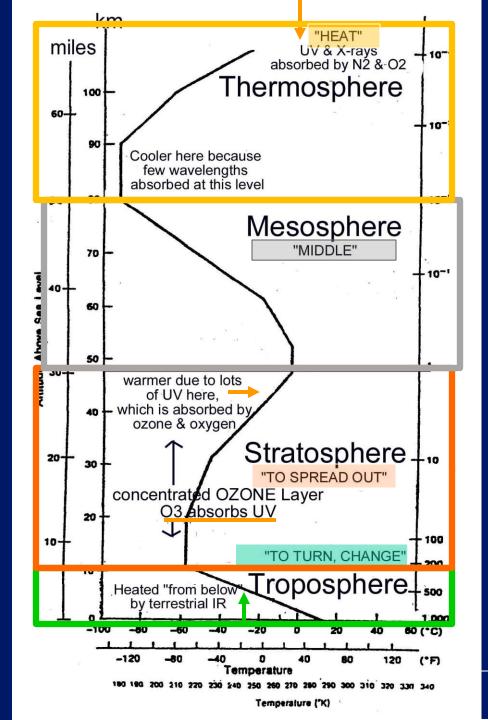
The changes in temperature with height are the result of:

differential absorption of shortwave (SW) & longwave (LW) radiation

by atmospheric GASES concentrated at various altitudes.

SUMMARY: Here's why these changes in temperature occur :

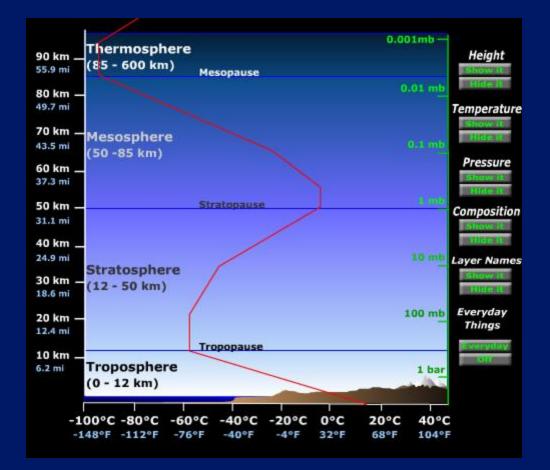
Start at the SURFACE \rightarrow



p 31

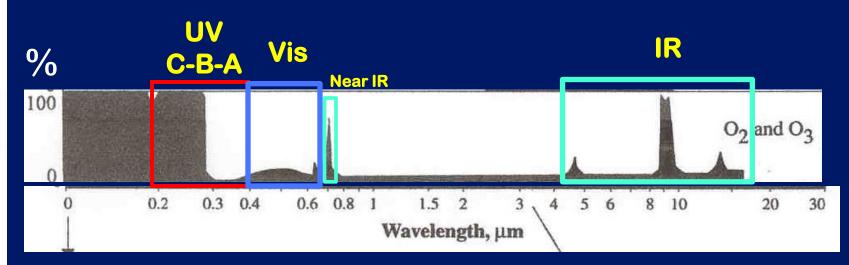
A nice online review ...

http://earthguide.ucsd.edu/earthguide/diagrams/atmosphere/index.html



<u>REVIEW</u>: The pattern of electromagnetic wavelengths that are absorbed & emitted by a particular atom (or combination of atoms)

is called its ABSORPTION SPECTRUM or its ABSORPTION CURVE

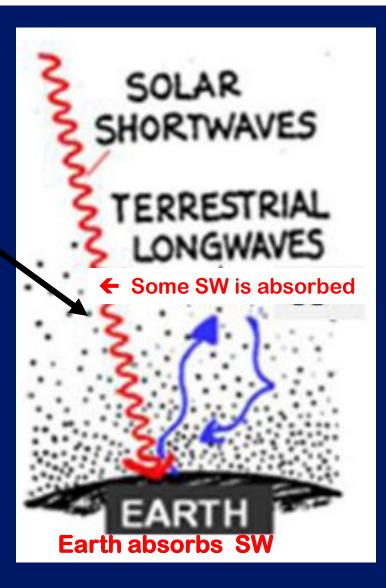


The Absorption curve for Ozone / Oxygen

There's one more thing to correct in our the depiction of incoming Solar . . .

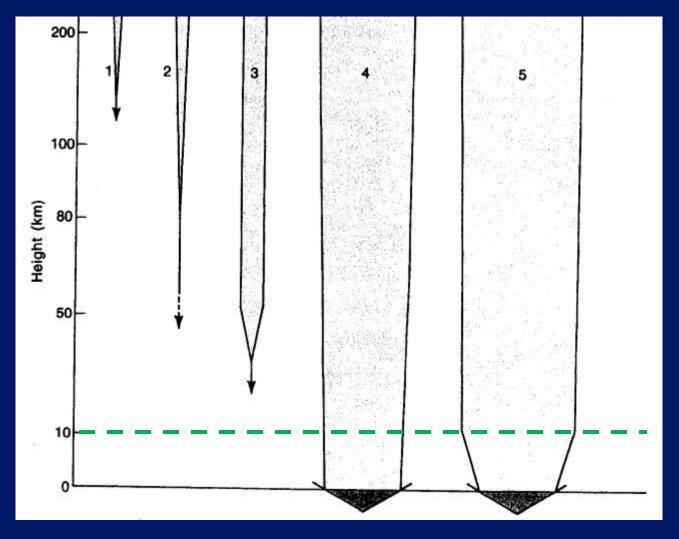
Some SW radiation gets absorbed on its way down to the surface!

(in addition to terrestrial LW (IR) radiation being absorbed in the GHE)



How incoming SOLAR radiation of different wavelengths gets TRANSMITTED or **ABSORBED** by different gases on its way to the Earth's surface

GROUP WORK: Study this box of info & answer Q1, Q2, & Q3



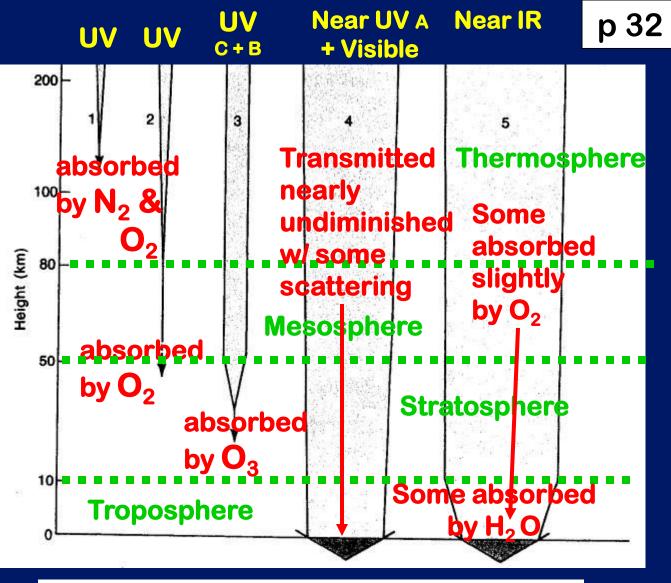
- 1. UV, $\lambda < 0.12 \,\mu$ m, absorbed by N₂ and O₂ in upper atmosphere
- UV, 0.12 µm ≤ λ < 0.18 µm absorbed by O₂
- 3. UV, 0.18 μ m $\leq \lambda < 0.34$, μ m absorbed by O₃ in ozone layer
- 4. Near UV and visible, 0.34 $\mu m \le \lambda <$ 0.7 μm transmitted nearly undiminished except for scattering
- 5. Near IR, 0.7 μ m $\leq \lambda <$ 3.0 μ m , absorbed slightly by O₂ and in troposphere by H₂O

Reminder: Ultraviolet radiation: UVC = 0.20 - 0.29 UVB = 0.29 - 0.32 UVA = 0.32 - 0.40 µm

UV rays < .32 µm very harmful to life on Earth arrows

1, 2 + 3

How incoming SOLAR radiation of different wavelengths gets TRANSMITTED or **ABSORBED** by different gases on its way to the Earth's surface



- 1. UV, $~\lambda$ < 0.12 μm , absorbed by N_2 and O_2 in upper atmosphere
- 2. UV, 0.12 μm ≤ λ < 0.18 μm absorbed by O₂
- 3. UV, 0.18 $\mu m~\leq~\lambda <$ 0.34, μm absorbed by O_3 in ozone layer
- 4. Near UV and visible, 0.34 $\mu m \le \lambda <$ 0.7 μm transmitted nearly undiminished except for scattering
- 5. Near IR, 0.7 μ m $\leq \lambda <$ 3.0 μ m , absorbed slightly by O₂ and in troposphere by H₂O

_Reminder: Ultraviolet radiation: UVC = 0.20 - 0.29 UVB = 0.29 - 0.32 UVA = 0.32 - 0.40 μm



THINK for 15 seconds TABLE CHAT for 15 seconds What's your most burning question?

ATMOSPHERIC COMPOSITION

Which gases? What concentration? Which ones are Greenhouse Gases (GHG)? Where do the GHG's come from?

Which GHG's are changing in concentration due to HUMAN ACTIVITIES? OVER THE WEEKEND STUDY THE TABLES ON Pages 33 & 34 to familiarize yourself with each of the GHG's

Then get ready for the "NAME THAT GAS!" Team Competition on Monday!



Participation Point Activity:

Get a blank index card from you group folder, put Name & Group # on it

Pick <u>one</u> of the following, reflect on what you learned today about it, & explain this in a short paragraph →

What we did today . . .

- 1) Reviewed the LEARNING PHILOSPHY of the course
- 2) Dug deeper into the pathways of incoming solar and outgoing terrestrial radiation and how the Greenhouse Effect is involved
- 3) Learned about Earth-Sun Relationships and what the an EQUINOX is – and how this causes the seasons and determines Earth's climate regions
- 4) <u>Sketched</u> the Vertical Structure of the Atmosphere and learned what causes it – including the role of certain gases in different layers

What to do: on the index card w/ name + Group # Pick <u>one</u> of the above , reflect on what you learned today about it, & explain this in a short paragraph

SEE YOU MONDAY



GO CATS!