### We're about to play:

PSUEDO-JEOPARDY!!!!!
(Aka "The Answer Is . . . ")

# And The ANSUERIS.

### Let's Meet Our Teams...

- **InfraRED Radiators**
- **TANGERINE Tasers**
- Mellow YELLOW Reflectors
- **GREEN House Gassers**
- BLUE Sky Diffusers
- UltraVIOLET Zappers

## Ready for a practice question?

**Energy Balance**  The circled symbol:

### What is...

- 1. Outgoing longwave radiation
- 2. Reflected infrared radiation

- 3. Libido
- 4. Albedo



Atmospheric Structure and Composition	Energy Balance	Matter & Thermo- dynamics	Laws of Motion & Radiation	Odds & Ends
<u>100</u>	<u>100</u>	<u>100</u>	<u>100</u>	<u>100</u>
<u>200</u>	<u>200</u>	<u>200</u>	<u>200</u>	<u>200</u>
<u>300</u>	<u>300</u>	<u>300</u>	<u>300</u>	<u>300</u>
<u>400</u>	<u>400</u>	<u>400</u>	<u>400</u>	<u>400</u>
<u>500</u>	<u>500</u>	<u>500</u>	<u>500</u>	<u>500</u>

### The gases: H2O and CO2.

#### What are...

- 1. The two most abundant gases.
- 2. The two most abundant Greenhouse gases.



- 3. The two most abundant <u>anthropogenically enhanced</u> Greenhouse gases.
- 4. The two gases that comprise 99% of the atmosphere

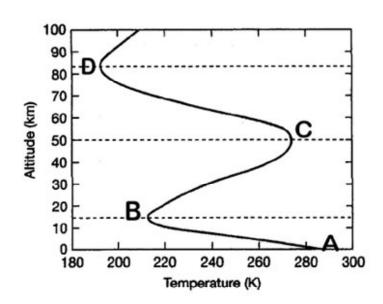
### The observation that "the atmosphere is heated from below" is most evident in this layer.

### What is...

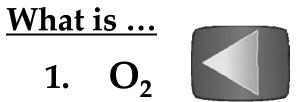
1. **Layer A - B** 



- 2. Layer B C
- 3. Layer C D
- 4. Layer D and above



### This gas is NOT a Greenhouse Gas.



2. O<sub>3</sub>

3. CH<sub>4</sub>

4. Freon-11 (a CFC)

### The residence time of CO2 gas molecules, once they get into the atmosphere.

#### What is...

- 1. ~10-12 years
- 2. ~50 years
- 3. ~100 years



4. ~ 500 years

### N2, N, O and O2 are effective absorbers of extremely harmful X-ray and UVC radiation in this layer.

#### What is...

- 1. Troposphere
- 2. Stratosphere
- 3. Mesosphere
- 4. Thermosphere



### Evaporation and transpiration are represented by this symbol.

### What is...

- 1. LW
- 2. SW
- 3. H
- 4. LE



5. **C** 

### The Greenhouse effect is represented by this symbol.

### What is...

1. This one: 
$$\overset{\text{sw}}{\checkmark} + \overset{\text{lw}}{\checkmark}$$

2. This one: 
$$\begin{cases} \sum_{l,w} + \sum_{l,w}^{l,w} \\ 1 \end{cases}$$

3. This one: 
$$\int_{-\infty}^{\infty} + \sum_{i=1}^{\infty}$$

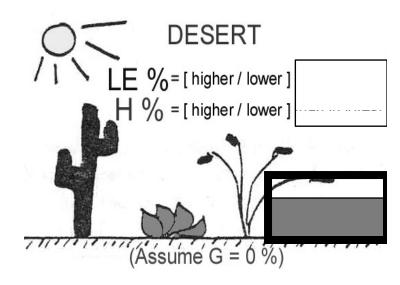
4. 
$$H + G$$

5. None of the above

### This will happen after a canal or reservoir is built in a desert.

#### Who is...



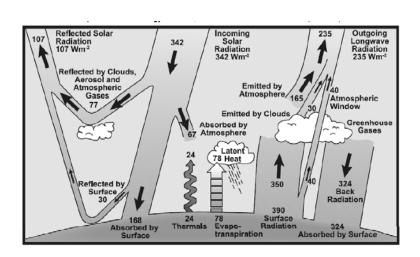


- More energy goes into H, making it hotter.
- 3. More energy goes into LE, making it hotter.
- 4. More energy goes into LE, making it cooler.





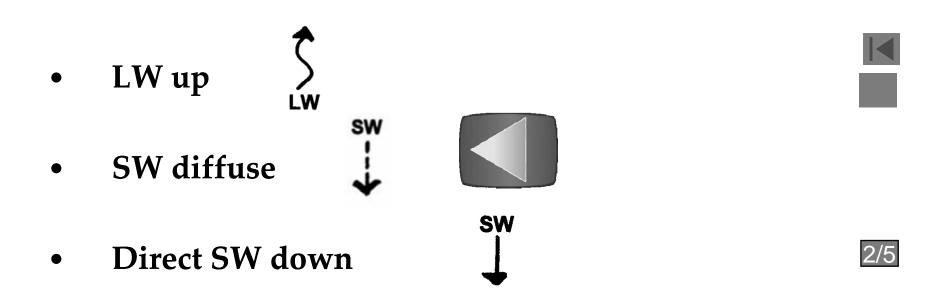
Of these 4 choices, THIS part of the energy balance and its pathways involves the MOST units of energy. (HINT: see p 64 in Class Notes What is...



- 1. UV energy absorbed by ozone in stratosphere
- 2. Solar energy reflected back to space by clouds, atmosphere, & surface combined.
- 3. IR energy radiated from the Earth's surface directly out to space
- 4. IR energy re-radiated to the Earth's surface after being absorbed in the atmosphere.



If the Earth had <u>NO atmosphere</u>, the amount of energy in THIS component of the Energy Balance would be minor compared to the other components



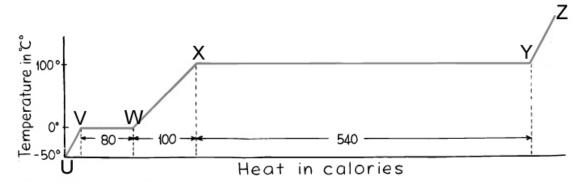
• Energy escaping thru the IR Atmospheric window

The reason why -- if Global Warming is occurring -- we should be able to detect it FIRST in LAND SURFACE temperatures rather than OCEAN SURFACE temperatures. What is...

1. The specific heat & heat capacity of WATER is higher than that of SOIL, hence water heats up more slowly than so

- 2. The specific heat & heat capacity of SOIL is higher that that of LAND, <u>hence soil heats up more slowly</u> than water.
- 3. The albedo of WATER is higher than that of SOIL, hence it will absorb more radiation

### The segments of this graph that represent LATENT energy (LE)



### What are:

- 1. U-V, W-X, and Y-Z
- 2. V-W and X-Y

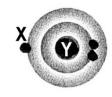


3. V-W and W-X

4. V- X and X - Z



## The number of positively charged protons the nucleaus of this neutral lithium atom contains. What is...



1. One

2. Two

3. Three

4. None - the nucleus contains photons, not protons!



Energy transfer by means of vibrational energy from one molecule to the next through a substance.

#### What is...

- 1. Convection
- 2. Conduction

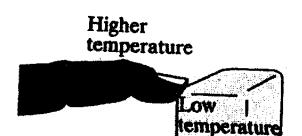


- 3. Radiation
- 4. Latent Energy

### How thermal energy will flow in this diagram, based on the 2nd Law of Thermodynamics

What is...

1. By means of CONVECTION

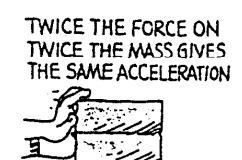


2. From the ICE CUBE to the FINGER

3. From the FINGER to the ICE CUBE



Based on the Newton's Law shown in this figure, the force (via consumption of gas) needed by either a Hummer or an Echo, to make <u>both</u> accelerate equally from a position at rest if the Hummer has 2 times the mass of the Echo.



#### What is...

1. The Hummer will need  $\frac{1}{2}$  as much force as the Echo

2. The Echo will need  $\frac{1}{2}$  as much force as the Hummer



3. The Hummer & Echo will need the same amount of force

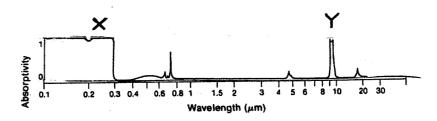


4. The Echo will need twice as much force as the Hummer

### The Radiation Laws that best explains why absorption curves exist. What is...

- 1. The hotter the body, the shorter the wavelength
- 2. Shorter electromagnetic wavelengths have higher intensity radiation than longer wavelengths **E = h c / λ**
- 3. The hotter the body, the (much) greater the amount of energy flux or radiation  $\boxed{E = \sigma 74}$
- 4. Some substances emit and absorb radiation at certain wavelengths only.

## The part of this O<sub>3</sub> absorption curve that is linked to OZONE'S absorption of harmful UV radiation in the stratosphere.



#### What is...

1. Part X of the absorption curve

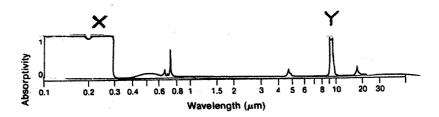


- 2. Part Y of the absorption curve
- 3. Both Parts X & Y working together
- 4. Neither X or Y this is NOT an absorption curve!





## The part of this O<sub>3</sub> absorption curve that is linked to OZONE operating as a GREENHOUSE GAS.



#### What is...

- 1. Part X of the absorption curve
- 2. Part Y of the absorption curve



- 3. Both Parts X & Y working together
- 4. Neither X or Y OZONE is NOT one of the Greenhouse Gases!





What ozone does in the troposphere vs. stratosphere.

#### What is...

1. Ozone absorbs IR in the troposphere (acting as a GHG) and absorbs harmful UV in the stratosphere (NOT acting as a GHG).



- 2. The Ozone Hole in the stratosphere allows more SW radiation to reach the Earth's surface and this leads <u>DIRECTLY</u> to an increase in the GREENHOUSE EFFECT.
- 3. BOTH of the above

### The wavelength range of infrared radiation.

### What is...

- 1. < 0.4 micrometers
- 2. > 0.7 micrometers

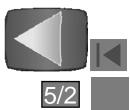


- 3. 400 700 nanometers
- 4. Longer wavelengths than microwaves



### The key factor that makes certain gases act as greenhouse gases! What is...

- 1. They are diatomic
- 2. They <u>absorb</u> shortwave radiation and <u>emit</u> longwave radiation
- 3. They easily <u>reflect</u> IR radiation back to the Earth's surface
- 4. They <u>absorb</u> and <u>emit</u> infrared radiation



## The tree ring core that represents a tree that is highly SENSITIVE to climate & good for crossdating:

What is...

2. This one:





The Newton's Law that is illustrated in this cartoon:

### What is...

- 1. The Inverse Square Law
- 2. The Law of Inertia



3. The Law of Entropy







### Quantum behavior of certain molecules (bending, rotation, vibrations)

#### What is...

- 1. Why photons leap to higher energy states
- 2. Behavior explained by Newton's Laws
- 3. The reason LE is not sensed as heat

4. The reason some gases are greenhouse gases and others are not.